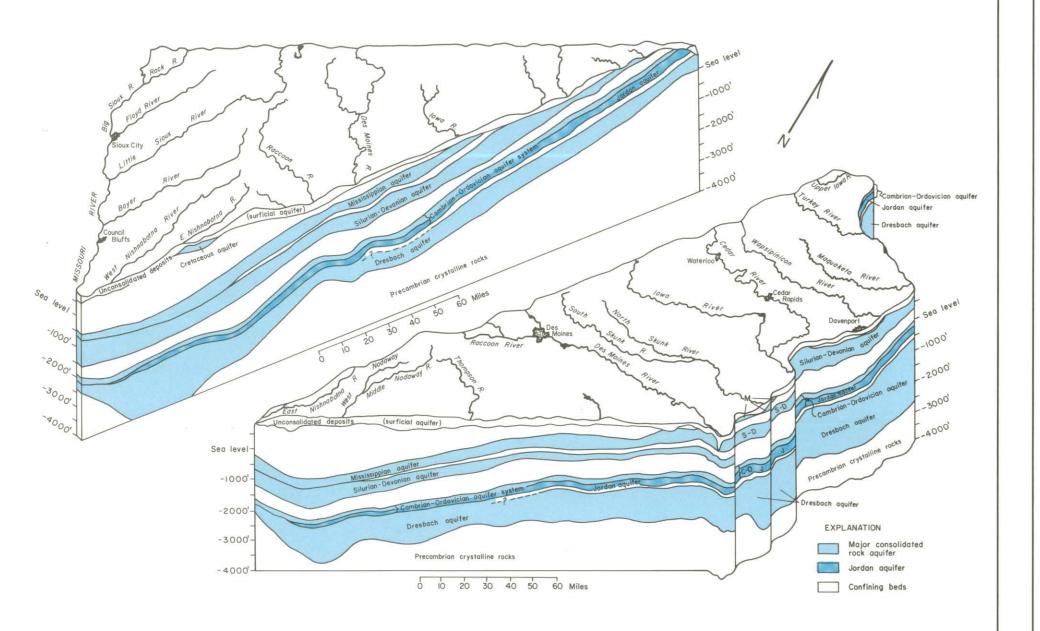
IOWA GEOLOGICAL SURVEY Prepared in cooperation with the STANLEY C. GRANT, DIRECTOR **United States Geological Survey**

INTRODUCTION

Water demand for all uses in lowa is increasing at an accelerated rate. Demand has increased from about 1,800 million gallons per day in 1955 to 3,500 million gallons per day in 1975 (MacKichan, 1957; Murray and Reeves, 1977). By the year 2020, water demand is expected to be eight times that in 1975 (Barnard and Dent, 1976). Historically, about 75 percent of the demand, excluding that required for power generation, has been met by withdrawals from the water-bearing zones (aquifers) in lowa's ground-water reservoir. Because this percentage is expected to

remain about the same, the anticipated future demands will re-

quire extensive withdrawals from the ground-water reservoir. The increasing stress on the ground-water system, which already is severely stressed in several places in the state, will create development and management problems that will require hydrologic information to solve. In order to provide this information, the Iowa Geological Survey in cooperation with the U.S. Geological Survey has instituted a series of investigations to define, describe, and evaluate the major aquifers in lowa's groundwater reservoir. Information from each investigation will be pre-



BLOCK DIAGRAM OF IOWA

IOWA'S GROUND-WATER RESERVOIR CONSISTS OF FIVE PRINCIPAL CONSOLIDATED ROCK AQUIFERS, AND THE UN-CONSOLIDATED DEPOSITS OF SAND AND GRAVEL THAT

OVERLIE THE BEDROCK. WIDESPREAD CONFINING BEDS

RETARD MOVEMENT OF THE WATER BETWEEN THE

The bottom of the ground-water reservoir is the Precambrian crystalline complex, which occurs at a depth of about 5,200 feet in southwestern lowa and rises to the surface in extreme northwestern lowa and to within 800 feet of the surface in northeastern lowa. Overlying these basement rocks is a succession of consolidated sedimentary strata of Paleozoic age that are dominantly sandstones and dolomites in the lower part, and shales, dolomites and limestones in the upper part. These strata have been downwarped into a broad trough, known as the lowa Basin. The surface of the dipping Paleozoic rocks was beveled by erosion, thereby exposing older Paleozoic strata in the northeastern and northwestern parts of the basin. This process created the extensive recharge areas of the Paleozoic aquifers in northeastern lowa and southern Minnesota. The beveled Paleozoic rocks in northwestern lowa are unconformably overlain by nearly flat-lying shales and sandstones of Cretaceous age. The

surfaces of both the Paleozoic and Cretaceous rocks have been modified considerably by pre-Pleistocene erosion, and nearly everywhere are mantled by glacial drift and loess of variable

The water-yielding consolidated rocks in the ground water reservoir are the porous, permeable sandstones of Cretaceous age and the sandstones and fissured, cavernous carbonates (limestones and dolomites) of Paleozoic age. One of the Paleozoic carbonate units, the Mississippian aquifer, was the subject of the first atlas published (Horick and Steinhilber, 1973). The present atlas is concerned with another Paleozoic unit — the Jordan aquifer, which is the lower and consistently most productive water-yielding unit of the Cambrian-Ordovician aquifer system. The Jordan aquifer is one of the most dependable sources of water supply for large capacity wells in lowa.

PURPOSE

The purpose of this atlas is to (1) define and describe the spatial relations and physical characteristics of the Jordan aquifer, (2) present information on the occurrence, availability, use, and chemical quality of water in the aquifer, and (3) define

and delineate changes in the potentiometric surface of the aquifer. The geohydrologic information has been divided into three subject headings that are presented on separate atlas sheets — geology, hydrology, and chemical quality.

ACKNOWLEDGMENTS

Acknowledgments are due to several people who assisted in acquiring data for this study. James F. Wiegand of the lowa Natural Resources Council was instrumental in obtaining the cooperation of water superintendents in maintaining water-level and pumpage records. Additional pumpage data of many Jordan aguifer wells were obtained from the Department of Environmental Quality files. The cooperation of several lowa drillers and engineering consultants, who submitted construction details, sample cuttings, water-level data, and pumping-test information,

cannot be stressed enough. The municipal water-plant operators also deserve praise for their cooperative attitude in keeping accurate records. Leon Steele and R. E. Hansen of the U.S. Geological Survey and Greg Ludvigson of the Iowa Geological Survey assisted in the collection of field data. R. J. Schliekelman provided information on radium-226 in Jordan aquifer well water and on treatment processes used for radium removal. Logan Kuiper of the Iowa Geological Survey critically reviewed the water-budget calculations.

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STRATIGRAPHIC AND HYDROLOGIC RELATIONS OF THE JORDAN AQUIFER

System	Rock-Stratigraphic Unit		Maximum Thickness	Physical Characteristics	Hydrologic Units	
ORDOVICIAN	Maquoketa Formation		300'	Mostly shale, grayish green, with dolomite beds in upper and lower part in east-central lowa; mostly dolomite, brown, with chert in north-central and western lowa; thin red shale with limonite or hematite pellets (Neda) at top.	Confining bed (locally water bearing in north-central lowa)	
	Galena Dolomite		230'	Dolomite, minor limestone, minor chert in lower half.	Water Bearing	
	Decorah Formation		170'	Limestone and dolomite, tan to brown; grayish-green and brown shales at top and base.	Confining Bed	
	Platteville Formation			Limestone, gray, and dolomite, brown, fossiliferous; shale, grayish-green at base (Glenwood). Sandstone, fine-to-medium grained above the shale in southeast lowa only.		
	St. Peter Sandstone		110'	Sandstone, coarse to fine, rounded and frosted grains, loosely cemented, minor green shale stringers.	Water Bearing	
	Prairie du Chien Formation	Willow River Dolomite Member	650'	Dolomite, sandy.	Confining Bed Later System	System
		Root Valley Sandstone Member		Sandstone, dolomitic	dan Aquifer	Cambrian-Ordovician Aquifer System
		Oneota Dolomite Member		Dolomite, crystalline, contains chert.		
CAMBRIAN	Trempealeau Group	Jordan Sandstone	145'	Sandstone, fine-to medium-grained, well-sorted and frosted grains; contains sandy dolomite beds in upper (Madison) and basal (Lodi) units.	Jorda	Cambr
	Tremp	St. Lawrence Dolomite	260'	Dolomite, coarsely crystalline, gray, silty, commonly containing glauconite	Confining	
	Franconia Sandstone		280'	Dolomitic siltstone, glauconitic shale, and glauconitic sandstone.	Bed	
	Dresbach Group	Galesville Sandstone	200'	Sandstone, medium- to coarse-grained, white to gray.	Water Bearing Confining Bed Water	uifer
		Eau Claire Sandstone	260'	Shale, silty, gray, fissile; siltstone, dolomitic; sandstone, fine-grained; some dolomite.		bach Aq
		Mt. Simon Sandstone	825'	Sandstone, medium-to coarse-grained with minor shale stringers.	Water Bearing	

The nomenclature and classification of rock units in this report are those of the lowa Geological Survey and do not necessarily coincide with those accepted by the U.S. Geological Survey.

DEFINITION OF THE AQUIFER

The Jordan aquifer, named for the Jordan Sandstone, is a part of the Cambrian-Ordovician aguifer system1 and consists of the Cambrian Jordan Sandstone, the principal water-yielding unit, and the Oneota Dolomite and Root Valley Sandstone Members of the overlying Ordovician Prairie du Chien Formation (see stratigraphic chart and geohydrologic section). These two members are included in the Jordan aquifer because they are always left open in "Jordan wells"

because it is not considered to be a major waterbearing unit. The extent, altitude, and thickness of only the Jordan Sandstone is mapped here, because this is the unit that must be penetrated for maximum well yields from the aquifer.

¹ The Cambrian-Ordovician aquifer system is defined as consisting of the Cambrian Jordan Sandstone and the Ordovician Prairie du Chien Formation and St. Peter Sandstone (Steinhilber and Horick, 1970).

PHYSICAL CHARACTERISTICS OF THE AQUIFER

As described above, the Jordan aquifer consists of two sandstone units separated by a dolomite unit. The total thickness of the aquifer ranges from about 400 to 450 feet in east-central and southeastern lowa to about 150 feet or less in western lowa. All units wedge out in extreme northwestern lowa, either because of nondeposition or erosion or both.

The lithology of the Jordan Sandstone is rather uniform. In the northern part of the state it is a white to buff, fine- to coarse-grained, quartzose sandstone that is loosely cemented. Dolomite beds occur in the upper and lower parts. In the southern part of the state the sandstone is similar but contains an increasing percentage of dolomite and is moderately to tightly cemented. The thickness of the formation ranges from a maximum of about 145 feet in northeastern lowa to about 30 feet in central and southwestern lowa, as shown on the map in the lower right hand corner.

The Oneota Member of the Prairie du Chien Formation is a drab-gray to buff, thick-bedded, crystalline, cherty dolomite that is locally vuggy, cavernous, and highly fractured. Its maximum thickness is about 235 feet in southeast lowa, from where it thins to about 190 feet in northeastern lowa and to 0 thickness in northwestern lowa.

and contribute substantial quantities of water in some

places. Although many "Jordan wells" are extended in-

to the upper part of the underlying St. Lawrence

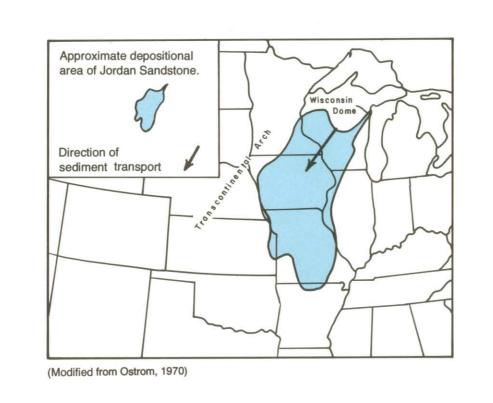
Dolomite, this unit is not included as part of the aquifer

The Root Valley Member of the Prairie du Chien Formation is very similar to the Jordan Sandstone, consisting of white to buff, fine- to medium-grained, quartzose sandstone; it is loosely cemented in northern lowa and becomes increasingly dolomitic and more tightly cemented to the south. The thickness of the Root Valley is extremely variable throughout the state. The maximum thickness occurs in central and southeastern lowa, where it is as thick as 110 feet; elsewhere it is as thin as 20 feet.

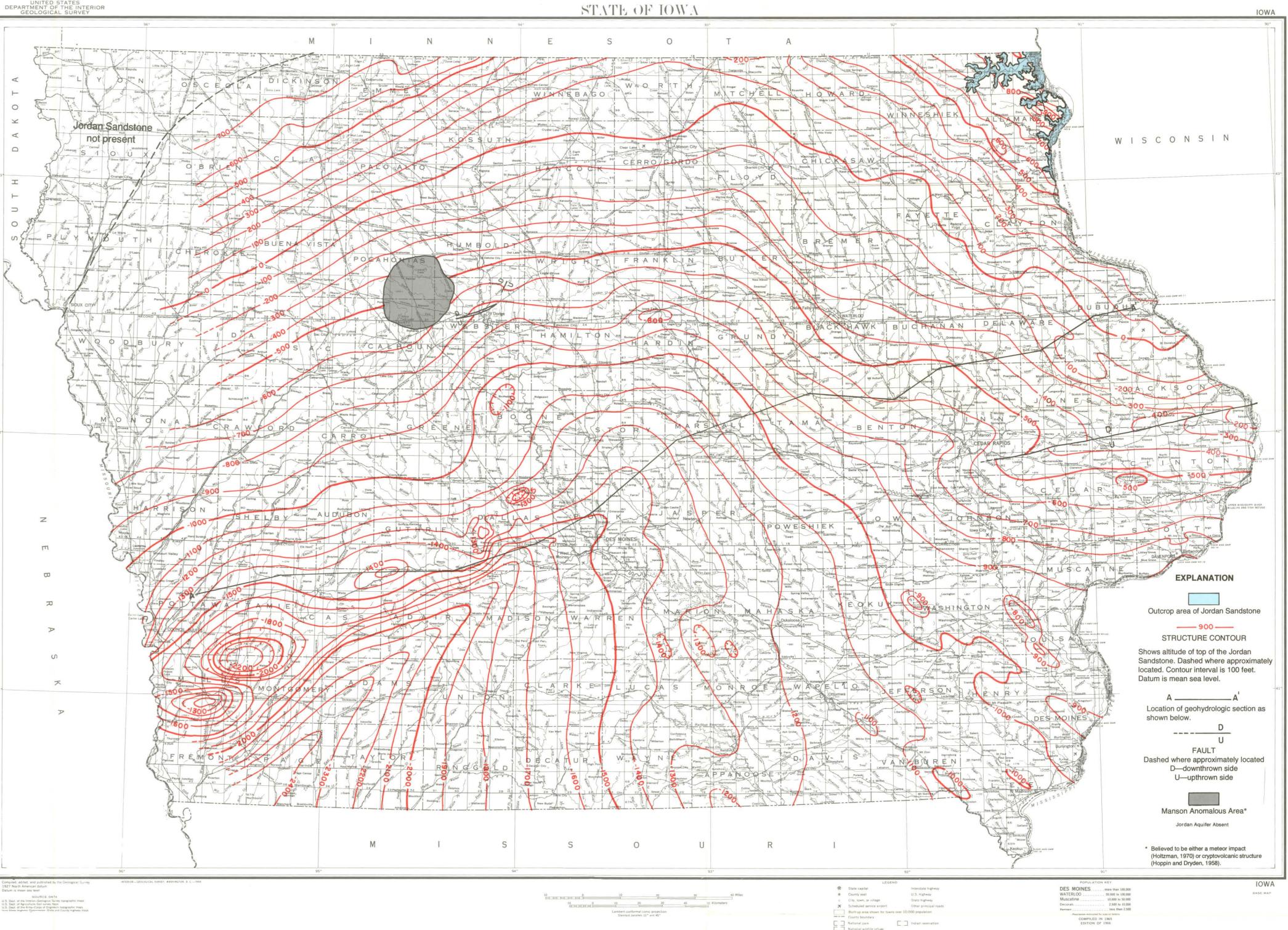
AREAL EXTENT

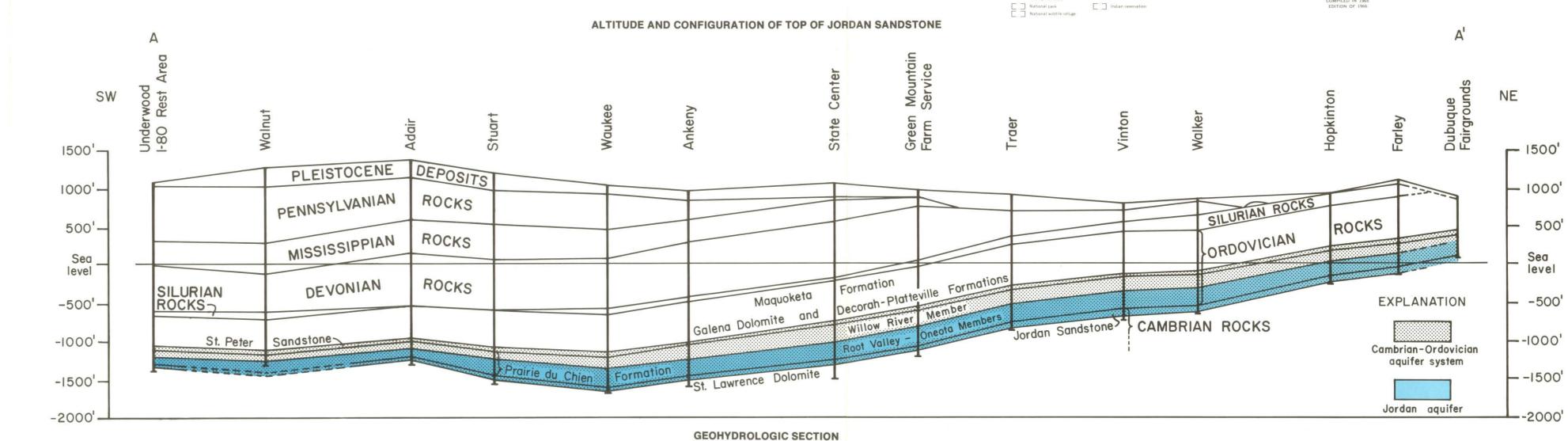
The lower Paleozoic sedimentary rocks of the Upper Mississippi Valley are considered to be cyclic deposits with each cycle consisting of a basal quartzose sandstone, overlain by reworked quartzose sandstone, clayey sandstone or shale, and capped by dolomite or limestone (Ostrom, 1970). Five cycles or partial cycles are identified in lowa rocks. The five quartzose sandstones are the Mount Simon, Galesville, and Jordan Sandstones, the Root Valley Member of the Prairie du Chien Formation, and the St. Peter Sandstone. The source of the sediments was the Wisconsin Dome which was periodically uplifted and eroded during Cambrian and Ordovician time. This caused fluctuations of sea level and deposition of the different types of sediment that characterize the cycles. The approximate depositional area of the Jordan Sandstone is shown on the map below. In Iowa the

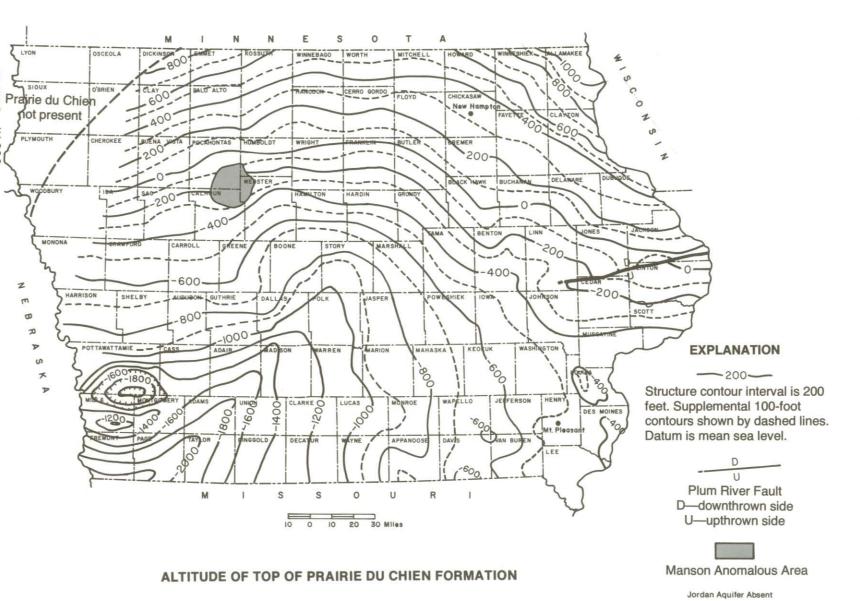
Jordan Sandstone underlies all of the state except the northwest corner and extreme southwest corner. The precise boundary of the formation in these areas is not known because it is covered by thick Cretaceous deposits in the northwest and thick Paleozoic units in the southwest. The only place the Jordan can be observed in surface exposures in Iowa is in Allamakee and Clayton Counties in the bluffs of the Mississippi River and in the vallevs of the Upper Iowa River and small tributaries of the Mississippi River. From this outcrop area, where its altitude is about 700 to 1000 feet above sea level, the formation dips southwesterly beneath younger Paleozoic strata to the deepest part of the lowa Basin where it is at an altitude of 2,400 feet below sea level. In the northwest part of the Iowa Basin the Jordan rises to 700 feet above sea level beneath the Cretaceous rocks in O'Brien and Dickinson Counties.



GEOLOGY



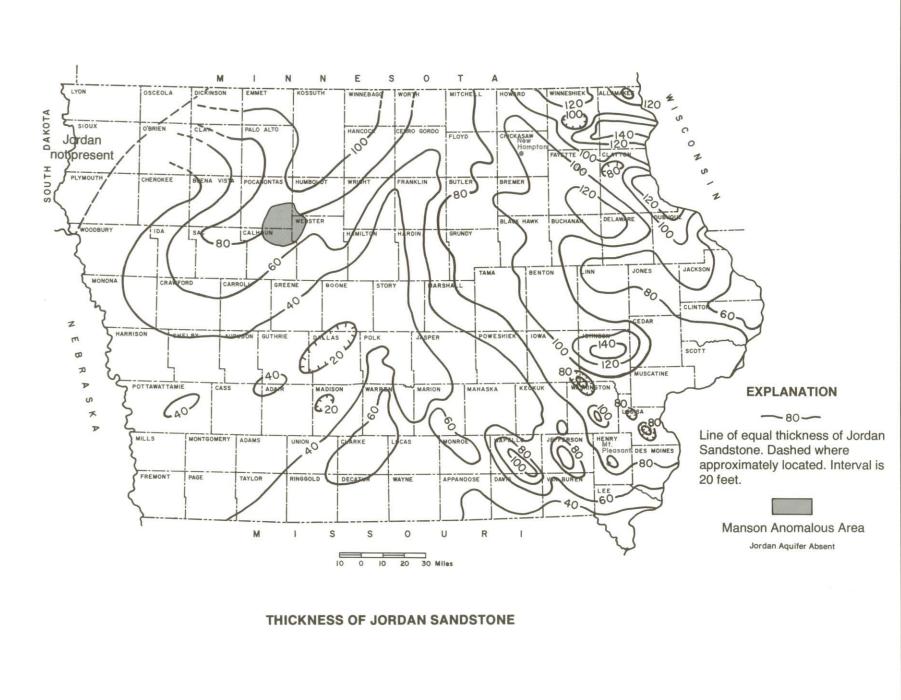




HOW TO CALCULATE WELL AND CASING DEPTHS

Useful information for predicting drilling depths and casing needs for Jordan aquifer wells can be obtained from the large map above showing the altitude of the top of the Jordan Sandstone, and the smaller maps opposite showing the altitude of the top of the Prairie du Chien Formation and the thickness of the Jordan Sandstone. Most modern Jordan aquifer wells are designed to be drilled about 25 feet into the St. Lawrence Dolomite and to be cased and grouted from the surface into the upper 50 to 100 feet of the Prairie du Chien to exclude all overlying water zones. Therefore, depending on whether the altitudes of the Prairie du Chien Formation and Jordan Sandstone are above or below sea level datum at a contemplated drilling site, the altitude from the maps can be subtracted or added to the surface altitude at the drilling location to obtain a reasonable estimate of the formation depths in feet below ground level. For example, at New Hampton, in Chickasaw County, the maps indicate the top of the Prairie du Chien lies at about 325 feet above sea level and the top of the Jordan Sandstone at about 25 feet below sea level. The thickness of the Jordan Sandstone is shown to be about 115 feet. With this information and assuming a surface altitude of 1,150 feet above sea level, the top of the Prairie du Chien would be expected at approximately 825 feet below ground level and the top of the Jordan Sandstone at about 1,175 feet below ground level. Thus, the bottom of the casing string probably would be set at about 875 feet, while the final well depth probably would be about 1300 feet. At Mt. Pleasant, in Henry County, based on a surface

altitude of 730 feet above sea level and the information from the maps, a Jordan aquifer well should be about 1,840 feet deep, with the bottom of the casing set at about 1,265 feet. All depth estimates obtained by using these maps are subject to some adjustment because of local structural and thickness variation of the strata.



MISCELLANEOUS MAP SERIES 6

SHEET 1 OF 3

JORDAN AQUIFER OF IOWA

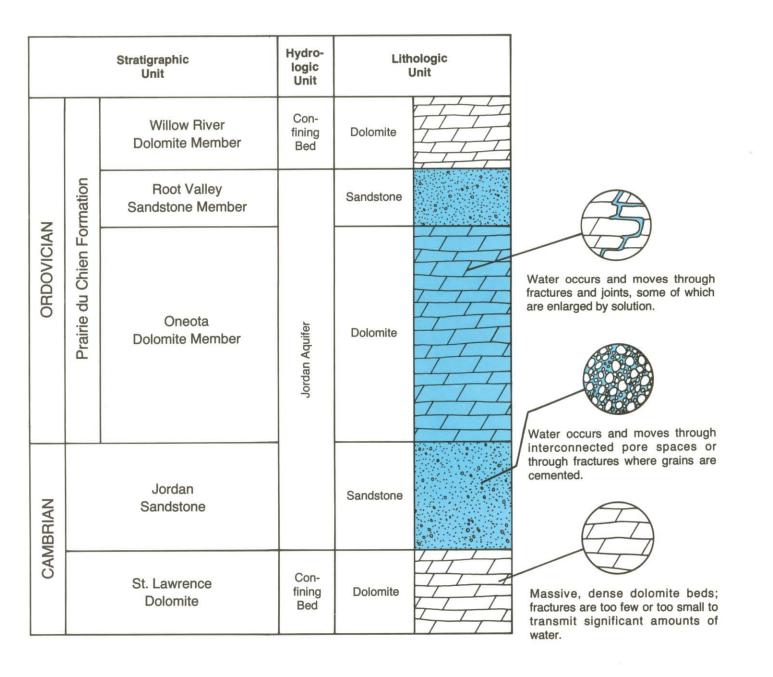
PAUL J. HORICK **Iowa Geological Survey** WALTER L. STEINHILBER **United States Geological Survey** Published by the STATE OF IOWA 1978

HOW WATER OCCURS IN THE JORDAN AQUIFER

WATER OCCURS IN THE JORDAN AQUIFER IN SANDSTONES AND CARBONATE ROCKS THAT HAVE BOTH PRIMARY AND SECONDARY POROSITY. From the time of deposition the sandstones had primary porosity except where the intergranular pore spaces later became filled with natural cementing substances. Secondary porosity was subsequently developed in some of the cemented zones by fracturing. The carbonate rocks, which are dense, crystalline dolomite, originally had a very low porosity. But as the sediments hardened and underwent structural stresses, openings developed along horizontal bedding planes and vertical fractures and joints. Groundwater containing dissolved carbon dioxide infiltrated these openings and dissolved some of the dolomite, thus enlarging the openings. Although the jointing in the dolomite probably follows a systematic pattern, the other openings generally are very irregular in distribution and size because of the variable solubility of the dolomite and the composition of the overlying materials through which the

water percolates. The sandstones probably have greater storage and transmission capabilities than do the dolomites, but locally the dolomites contain large fractures or solution caverns and will yield more water than the sandstones. THE WATER IN THE JORDAN AQUIFER IS UNDER ARTESIAN CONDITIONS; THAT IS, THE WATER IS UNDER PRESSURE AND RISES IN WELLS ABOVE THE TOP OF THE AQUIFER OR ABOVE THE LEVEL AT WHICH WATER WAS FIRST ENCOUNTERED IN THE AQUIFER. Originally the pressure was sufficient to cause natural flows from wells in the valleys of the large rivers of eastern and central lowa. The pressure surface has declined gradually over the years as more and more wells have tapped the aquifer for water supply. At major pumping centers the withdrawal rates are so high that water is being withdrawn from the aquifer more rapidly than it is being recharged, for example, at Des

Moines, Ft. Dodge, and Cedar Rapids.



ESTIMATED WATER BUDGET FOR THE REGIONAL FLOW SYSTEM OF THE AQUIFER

Water enters the Jordan aquifer principally by infiltration of the precipitation that falls on the intake area of northern lowa and southern Minnesota. The infiltrating water moves through Pleistocene drift directly into the aquifer in the northernmost part of the intake area. In the rest of the intake area the water moves through Pleistocene drift, sandstones and shales of Cretaceous age, and carbonate rocks of Devonian and Ordovician age before entering the aquifer. From the intake or recharge area, the water moves laterally southeastward by subsurface flow toward Illinois and Missouri. Discharge is into the Mississippi River north of the Jackson-Clinton County line and into the Illinois Basin via western Illinois and northern Missouri south of the line. During its movement from recharge to discharge areas, large quantities of water are extracted from the system by pumping wells, and significant amounts are added to the system by leakage from overlying aquifers in central and eastern lowa. The amounts of water added to and removed from the system a water budget of the system — have been calculated and the results portrayed on the map below.

The budget of the regional flow system of the aquifer can be expressed by the following equation (Lohman, 1972, p. 63).

 $R + \Delta R = D + \Delta D + q + S \Delta h$

R = recharge rate, ΔR = change in recharge rate; in this example $\Delta R = \Delta R_1 + \Delta R_2$, where ΔR_1 is leakage from overlying or underlying aquifers, and ΔR_2 is increased underflow from recharge area by

where the parameters, per unit area, are:

gradient steepening, D = natural discharge rate,

 ΔD = change in natural discharge rate,

q = rate of withdrawal from wells, S Δh = rate of change in storage.

In this example, leakage (ΔR_1) will be left as the residual, because data are available to estimate the other parameters. Recharge (R) and discharge (D) — Prior to withdrawals by wells, the system was in dynamic equilibrium. Recharge equaled discharge, and interaquifer leakage was minimal. Both recharge and discharge can be determined from the predevelopment potentiometric and transmissivity maps by using Darcy's Law to calculate the flow between two adjacent contour lines. The calculated values for recharge and dis-

charge are about 6.5 to 7.5 billion gallons per year. Change in discharge (ΔD) — An estimated 1 to 1.5 billion gallons per year of water that formerly discharged into northern Missouri has been diverted to the pumpage centers in the central part of the state. The gradient in most of eastern lowa has been decreased approximately 20 percent, so that the present discharge along the eastern lowa border is believed to be about 1 to 1.5 billion gallons per year less than during predevelopment time in that area. Thus, the total ΔD is about 2 to 3 billion gallons per year.

Change in recharge (ΔR_2) — The gradient in the recharge area of northern lowa, shown on the 1974-75 potentiometric map, has increased since predevelopment time by about 42 percent. Thus, the underflow from the main recharge area, which is proportional to the gradient, has been increased 42 percent or 2.7 to 3.2 billion gallons per year. Total pumpage (q) in the area of the regional flow system

in lowa equals 16.5 to 17.5 billion gallons per year.

Change in storage \(\Delta t \) - The change in state-wide aquifer storage since pumping began is about 175 to 200 billion gallons, which was determined from the potentiometric-change map and by assuming a state-wide storage coefficient of 2.5x10-4. Because about 30 percent of the change in the storage is attributed to withdrawals during the last 10 years (as indicated on the historical pumpage table), the annual average change (decrease) in storage attributed to pumping is about 5.3 to 6 billion gallons per year. This amount is approximately equivalent to an average state-wide head loss of about 1.8 to 2.4 feet per year as shown in the ad-

jacent hydrographs. Therefore, the average leakage to the aquifer, which is attributed to increased interaquifer head differentials, is estimated as:

 $\Delta R_1 = D - \Delta D + q - S \Delta h - R - \Delta R_2$

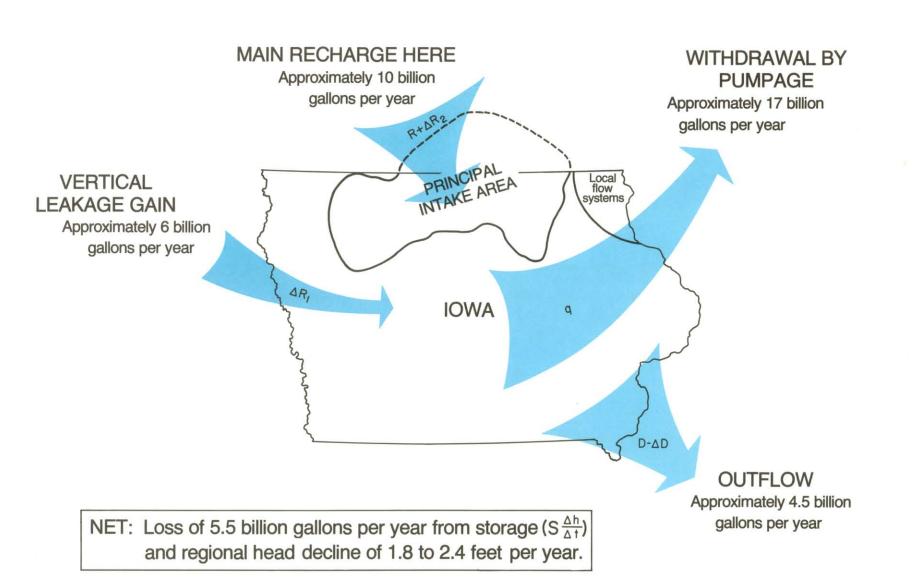
 $\Delta R_1 = 7 - 2.5 + 17 - 5.5 - 7 - 3 = 6$ billion gallons per

and: $R + \Delta R_1 + \Delta R_2 = D + \Delta D + q - S \Delta n$

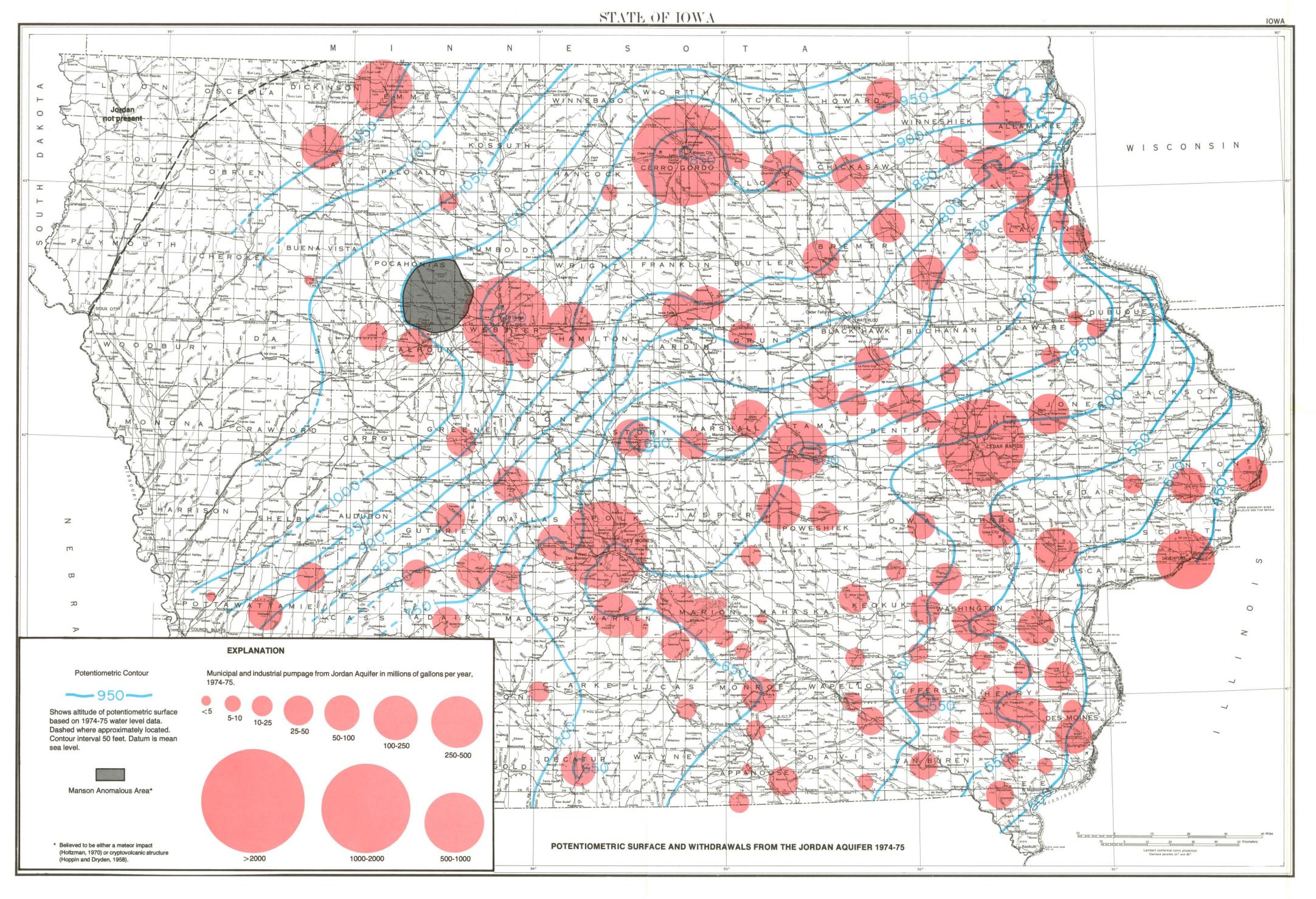
7 + 6 + 3 = 7 - 2.5 + 17 - 5.521.5 5.5

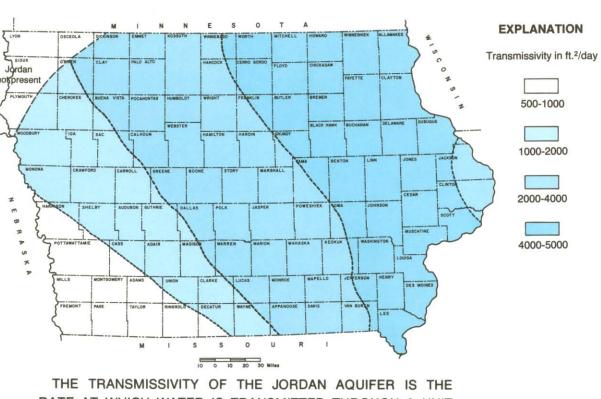
thus: (recharge) = (discharge) - (storage loss)

Because storage is being depleted, it is important to estimate the amount of water stored in the aquifer. Storage can be considered either as artesian storage — the amount of water that is released when the potentiometric surface is lowered by pumping, or as total storage — the amount of water contained in the aquifer openings (the porosity of the aquifer). The former is a more practical measure, because, except for some northeastern counties, dewatering of this deeply buried aguifer isn't likely in the foreseeable future. A meaningful value for artesian storage is the amount of water stored per foot of artesian head, which is about 2.4 billion gallons. Thus, if the potentiometric surface were lowered 100 feet throughout the State, excluding western and southwestern lowa, the amount of water derived from artesian storage would be about 240 billion gallons.



HYDROLOGY





500-1000

1000-2000

2000-4000

4000-5000

HYDRAULIC CHARACTERISTICS OF THE AQUIFER

usually high specific capacities occur, the transmissivities

may be somewhat higher than shown on the map. The

storage coefficient of the aquifer, based on a few aquifer tests

in the northern and eastern parts of the state, is estimated to

average about 2.5 x 10⁻⁴ (the value in the southwestern part

may be somewhat lower). The value of 2.5 x 10⁻⁴ was used in

calculating artesian storage in the aquifer and in calculating

John Morrell and Co. Wells

beginning in 1973 when pumping was stopped.

Le Claire Town Well

vious year's water level.

Hydrograph of well tapping the Jordan aguifer at Ottumwa

in Wapello County, showing head loss over the years result-

ing from industrial pumpage. Observe the sharp recovery

Hydrograph of abandoned well at Le Claire in Scott County,

showing a regional decline of the potentiometric surface

and annual fluctuations caused by industrial withdrawals

from wells in the Quad Cities area. Note that subsequent re-

coveries during the winter months do not regain the pre-

RATE AT WHICH WATER IS TRANSMITTED THROUGH A UNIT WIDTH OF THE AQUIFER UNDER A UNIT HYDRAULIC GRADIENT. It is expressed in square feet per day.

The transmissivity of the aquifer, shown on the above map,

is estimated to range from about 500 ft²/day in the western

part of the state to about 5000 ft²/day in the northeastern part.

The boundaries shown are highly generalized. These regional

estimates are based on state-wide specific capacity data and

on several aquifer tests in the eastern part of the state

(Hansen, 1970; and Hershey and others, 1970). At a few

American Crystal Sugar Co. Well

Mt. Pleasant City Well No. 4

tapping this source.

The effect of industrial and municipal pumpage on the static

head of the water level is shown in the hydrograph of a well

penetrating the Jordan aguifer at Mason City in Cerro Gordo

County. The sharp recovery in the period 1975-76 is attribut-

ed to shutdown of one of the major industrial consumers

Hydrograph indicating general decline of the water level in

the Jordan aguifer at Mt. Pleasant in Henry County. For the

30-year period of record, the total head loss is about 54 feet.

localities in southeastern lowa and in Polk County, where un-

Location of earliest wells drilled to the aquifer in a given area. Number refers to altitude of water level in well. Potentiometric contour Contour interval is 50 feet. Dashed Datum is mean sea level. Recharge in Minnesota and Iowa; discharge to Mississippi River in M I S S O U R I PRE-DEVELOPMENT POTENTIOMETRIC SURFACE OF JORDAN AQUIFER BASED ON EARLIEST WATER LEVEL RECORDS

Dubuque and Jackson counties, and to Illinois and Missouri by underflow. Recharge locally by precipitation and vertical seepage. Discharge to Upper Iowa and Mississippi Rivers. AVAILABLE FROM DEEP WELLS DATING BACK TO 1896 (NORTON, 1912). This map may be compared with the latest potentiometric map above prepared from 1974-75 data to estimate the head loss that has occurred in the intervening years and as shown on the map at right.

THE POTENTIOMETRIC SURFACE — a record of decline SINCE THE FIRST WELLS DRILLED TO THE JORDAN AQUIFER BEGAN WITHDRAWING WATER IN THE LATE 1800's, THE POTENTIOMETRIC SURFACE IN IOWA HAS DECLINED ABOUT 50 TO 150 FEET REGIONALLY AND AS MUCH AS 175 TO 200 FEET AT THE MAJOR PUMPING CENTERS. These recessions are shown on the small map above, which shows the changes between the predevelopment and 1975 potentiometric maps. Water-level trends at some pumping centers, shown on the hydrographs, indicate

that the water levels are declining at an average rate of 1.8 to

EXPLANATION

Regional flow system

Local flow system

2.4 feet per year. These declines will increase as long as the pumpage rates increase as indicated by the adjacent historical pumpage table. The water-level declines at most major pumping centers are about 4 to 6 feet per year. The large and deep drawdown cones created by these heavy withdrawals can be mitigated somewhat by spacing new wells farther apart in order to reduce mutual interference effects. A tool to calculate interference effects and thereby determine optimum

M I S S O U R I

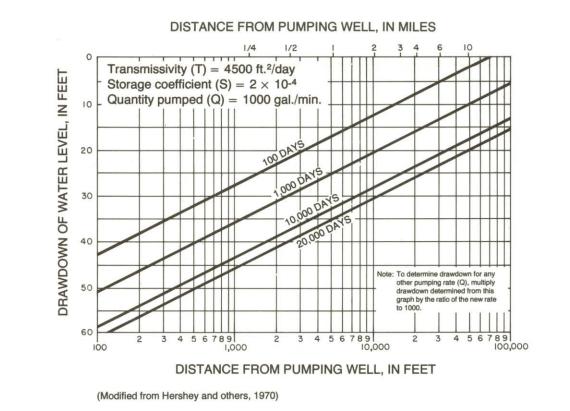
pumping centers the head loss is somewhat greater.

10 0 10 20 30 MHes

AREAL RECESSION OF POTENTIOMETRIC SURFACE OF

JORDAN AQUIFER. This map shows the regional decline. At the

well spacing is the distance-drawdown graph shown below.



A USEFUL TOOL IN DETERMINATION OF THE WATER LEVEL DRAWDOWNS THAT WILL OCCUR AT SPECIFIC DISTANCES FROM A PUMPING WELL FOR VARIOUS PERIODS OF TIME IS THE DISTANCE-DRAWDOWN GRAPH. A reproduction of one such graph is shown above (Hershey and others, 1970). The graph will assist in selecting the optimum spacing between Jordan aquifer wells to prevent

serious interference effects. Drawdowns determined for new wells must be superimposed on the existing potentiometric map. The graph is considered valid for localities in the eastern half of the State based on pumping tests at Mason City and Cedar Rapids. Other graphs based on different transmissivities will have to be constructed for other

WITHDRAWALS FROM THE AQUIFER

ANNUAL WITHDRAWAL OF WATER FROM THE JORDAN AQUIFER IN IOWA DURING 1974-75 WAS ABOUT 18 TO 19 BILLION GALLONS. OF THIS QUANTITY, ABOUT 12 TO 13 BILLION GALLONS OR ABOUT 66 PERCENT WAS WITHDRAWN BY MUNICIPALITIES AND ABOUT 6 BILLION GALLONS OR ABOUT 34 PERCENT WAS WITHDRAWN BY IN-DUSTRIES. All known pumping centers are shown on the adjacent map; the principal pumping centers, where about 85 percent of the water was withdrawn, are itemized in the table

Withdrawals from the Jordan aquifer are expected to increase in the future. Limited historical pumpage records, shown on the bottommost table, indicate an increase in

withdrawals by many municipalities; this trend is expected to continue or even to accelerate.

QUANTITY OF WATER ANNUALLY PUMPED BY

INDUSTRIES FROM THE JORDAN AQUIFER, 1974-75

below. Industrial pumpage is itemized in the other table

below. An unknown, but probably small amount of domestic

pumpage occurs in and near the outcrop area in northeastern

QUANTITY OF WATER WITHDRAWN BY MUNICIPALITIES AND INDUSTRIES IN MILLION GALLONS PER YEAR AT MAJOR PUMPING CENTERS, 1974-75a

Perry DeWitt

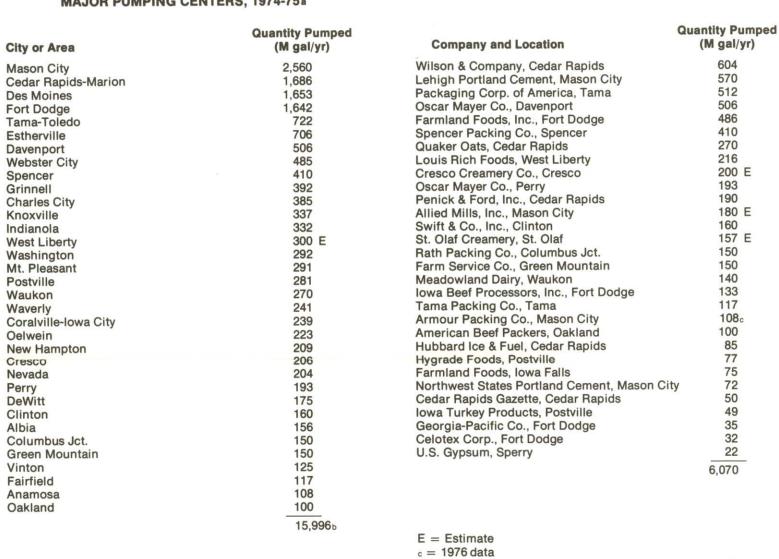
Clinton

EXPLANATION

Line of equal change in water level

(late 1800's) to 1975. Dashed

where approximately located.



YIELDS, SPECIFIC CAPACITIES, AND **DEVELOPMENT TECHNIQUES**

PENDABLE GROUND-WATER SOURCES FOR LARGE CAPACITY WELLS IN IOWA. Generally yields from the Jordan aquifer range from several hundred gallons a minute (gal/min) to 1,000 gal/min, and occasionally 2,000 gal/min in the northeastern two-thirds of the State. However, yields vary somewhat dependent upon the amount of cementation of the sandstones and the presence or absence of fractures in the dolomites. The construction method and the extent and technique of well development also have a bearing on the yield. The Jordan Sandstone is the principal water-yielding unit, but some unusually high capacity wells in south-central lowa (Polk, Dallas, Marion, Warren, Madison, and Guthrie Counties) and parts of southeastern Iowa (Ottumwa-Eldon, Donnellson, and the Tama-Grinnell areas) may obtain most, or at least much, of their water from fractures in the dolomites of the lower part of the Prairie du Chien Formation. The specific capacities of these wells may range as high as 20 to 40 (gal/min)/ft. of drawdown. Elsewhere in the eastern half of the State and parts of northwestern Iowa (Rockwell City-Ft. Dodge-Webster City, Spencer-Estherville, and Storm Lake-Odebolt areas) the aquifer generally will produce at least 4 to 5 (gal/min)/ft. of drawdown and range up to as much as 10 to 15 (gal/min)/ft. In southwestern lowa, where the sandstone is thin and the aquifer is deeply buried, specific capacities

THE JORDAN AQUIFER IS ONE OF THE MOST DE-

a Includes industrial pumpage, which is shown in adjacent table.

b Represents about 85% of total withdrawal from Jordan aquifer.

generally are less than 3 (gal/min)/ft. of drawdown. Proper well construction and development are important in obtaining maximum yields from wells that tap the Jordan aquifer. The most successful wells usually have a bottom hole diameter of at least 8 inches and are cased from the surface into the upper part of the Prairie du Chien Formation with the pipe grouted in place with neat

cement for its full length.

Development methods consist of treating the well with large quantities (4,000-8,000 gallons is common) of concentrated, inhibited hydrochloric acid, usually 20° Baume, in one or two stages, using a large volume of water to force the acid into the formation. The acid-water mixture may also be surged with a pressure pump. Some contractors use explosives to blow the acid out into the formation. Dynamiting the water-bearing interval or shooting with shaped charges are other development methods less

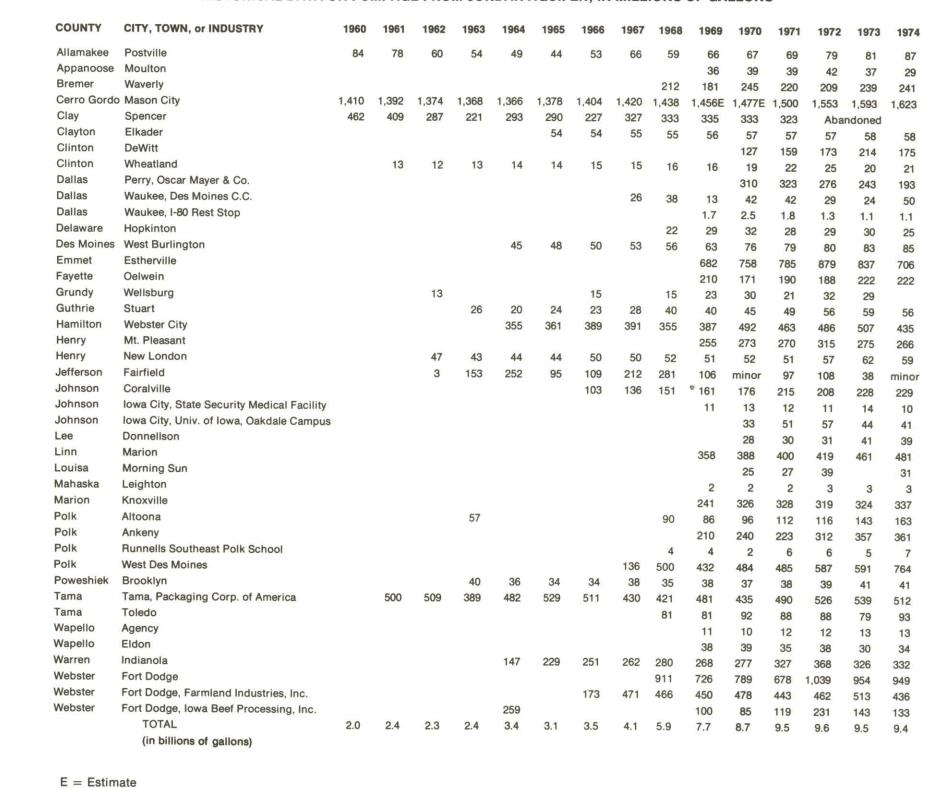
Some of these industries have additional sources of water, thus,

the figures shown do not necessarily indicate total water use by

These development methods exhibit a wide range of success. Increases of as much as 1,000 percent in specific capacity have been obtained, but the average increase in 23 wells from which data on development were collected was about 375 percent. In six of the wells the increase in specific capacity was less than 100 percent. This information indicates that the added expense of development usually is justified. The comparatively poor improvement in some wells is attributed to several factors such as failure to clean the hole before placement of the acid, insufficient acid used or acid of insufficient strength, acidizing without pressure, using only one treatment when two or three stages may be needed, failure to clean out the sand accumulated in the well from the development work, and shooting or blasting the well at the wrong depth, or with weak charges.

A technique of drilling and developing the Jordan Sandstone by injection of air under high pressure with a large volume of water holds considerable promise for constructing high capacity Jordan aquifer wells. With this method the well can be completed in one step without going back into the hole with the more conventional acid or blasting treatment. Very good results have been obtained with this method in the few places it has been

HISTORICAL DATA ON PUMPAGE FROM JORDAN AQUIFER, IN MILLIONS OF GALLONS



SUMMARY OF HYDROLOGY

The Jordan aquifer, which is widespread across lowa, is the principal source for large quantities of water (1,000-2,000 gal/min) for municipal and industrial use from wells penetrating the consolidated rocks. The aquifer occurs at a depth of about 1,000 to 1,500 feet in the northern and northeastern counties and at about 2,500 to 3,000 feet in the central and south-central counties. The artesian head of the aquifer has lowered as much as 175 to 200 feet in the major pumping centers, causing steep and extensive drawdown cones since the original wells were drilled 75 to 85 years ago. The annual head loss in

most of the State caused by the current imbalance between recharge and discharge is about 1.8 to 2.4 feet. Mining of water from the Jordan aquifer is expected to continue for the foreseeable future. However, there is no immediate danger of dewatering the aquifer, because of the large amount of water stored in the aquifer and because leakage from overlying rocks supplies a large percentage of the water withdrawn. Problems of local overdevelopment and serious interference may occur at the major pumping centers unless new wells are properly

JORDAN AQUIFER OF IOWA

PAUL J. HORICK **Iowa Geological Survey** WALTER L. STEINHILBER **United States Geological Survey** Published by the STATE OF IOWA 1978

IOWA GEOLOGICAL SURVEY Prepared in cooperation with the MISCELLANEOUS MAP SERIES 6 **United States Geological Survey** STANLEY C. GRANT, DIRECTOR SHEET 3 OF 3

EXPLANATION

___500___

maximum level proposed for community

solids. Interval is 250 mg/L.

in those areas where the fluoride

sulfate concentration exceeds 600 mg/L.

EXPLANATION

___50___

Line of equal temperature.

Interval is 5°F.

EXPLANATION

1 Estherville No. 9

3 Ossian No. 2

5 West Bend

8 Storm Lake

9 Hampton No. 3

10 Waverly No. 2

-

11 Arlington No. 2

12 Garnavillo No. 2

13 Rockwell City No. 4

14 Webster City No. 5

15 Wellsburg No. 1

16 Farley No. 3

18 Nevada No. 2

17 Jefferson

19 Toledo

20 Vinton No. 2

21 Anamosa

22 Altoona No. 2

23 Brooklyn No. 5

24 Coralville

25 DeWitt No. 5

26 LeClaire No. 3

29 Adair

27 D.O.T. I-80 Reststop

30 Indianola No. 10

31 Knoxville No. 2

32 What Cheer

34 Murray

35 Russell

38 Centerville

39 Keosaugua

31 Washington No. 7

37 Mt. Pleasant No. 4

4 Waukon No. 4

6 Mason City No. 14

7 Charles City No. 6

water supplies.

for most uses.

QUALITY OF WATER

SIGNIFICANCE OF CHEMICAL CONSTITUENTS AND PROPERTIES OF WATER

The chemical constituents and properties shown in the table below are the ones that are usually determined in routine inorganic and radiochemical water analyses. They form the basis for the chemical characteristics maps and text on this atlas sheet. The contaminant levels indicated and comments regarding the significance and acceptability of water are based on the U.S. Environmental Protection Agency's National Interim Primary Drinking Water Regulations and on the proposed Secondary Drinking Water Regulations. The primary regulations, which became effective in June 1977, established maximum contaminant levels for chemical constituents and other properties of public water supplies that affect the health of consumers. Because lowa has accepted the primary regulations, these are now enforced by the Iowa Department of Environmental Quality (DEQ). The proposed secondary regulations will establish maximum contaminant levels for constituents that affect the esthetic quality of water. When promulgated, these secondary regulations will be accepted by the DEQ as lowa's esthetic quality standard for public water

The maximum contaminant level of some constituents that affect the esthetic quality of water are exceeded in many public

supplies withdrawn from the Jordan aquifer. Many of these supplies have been used for many years without obvious ill efavailable to the community; any improvement in quality would require demineralization of the water.

Standards for uses other than drinking water are usually different than those listed below. The quality requirements for industries, for example, would depend on how the water was used and therefore are quite varied. The concentrations of some constituents and properties of water from the aquifer are shown on maps so the users can decide if the water meets their particular requirements.

The nonradioactive chemical constituents discussed in this atlas are expressed as ions in concentrations of milligrams per liter (mg/L). One mg/L is 1/1000th of a gram (milligram) of an ion in a volume of 1000 cubic centimeters (one liter) of water. An approximate weight to weight ratio would be one gram of the ion in 1,000,000 grams of water, or one part per million (ppm). Radium concentration and gross alpha activity are reported in picocuries per liter (pCi/L). One picocurie is the quantity of radioactive material producing 2.22 nuclear

Constituent or	Maximum contaminant levels in community water supplies ¹		Significance and comments		
property	Primary Proposed secondary regulations				
Iron (Fe)		0.3 mg/L	Iron concentrations of 0.3 mg/L or more are likely to be troublesome as iron cal cause red-water staining of plumbing fixtures and clothing and plug well screen and water pipes. High concentrations affect the color and taste of beverages.		
Silica (Si0 ₂) NA ²		A ²	Silica contributes to formation of boiler scale, forms deposits on blades of steaturbines, but is not physiologically or agriculturally significant. Most water from the Jordan aquifer has concentrations of 5 to 30 mg/L.		
рН	H NA		The pH expresses the negative logarithm of the hydrogen-ion concentration a indicates whether water is acid or alkaline. A pH of 7.0 indicates a neutral so tion, less than 7.0 is acid, more than 7.0 is alkaline. Most supplies from Jordan aquifer have pH values ranging between 7.1 and 7.9 and, therefore, slightly alkaline.		
Specific conductance (micromhos/cm. at 25° C)	NA		Specific conductance refers to the ability of the water to conduct an electric current and generally increases in a direct proportion to the presence of ionispecies in solution. There is a consistent enough relationship between the dissolved solids and specific conductance of water from the Jordan aquifer that the dissolved solids can be estimated by multiplying the specific conductance be 0.67.		
Dissolved solids (residue on evaporation at 180° F)		500 mg/L	Dissolved solids indicate the concentration of dissolved minerals in the water and is a measure of the water's suitability for many uses. Concentrations of over 2,000 mg/L, which in lowar implies a high concentration of sulfate, may have detectable taste and a laxative effect on people not acclimated to the water Many water supplies, particularly in western and southern lowar, contain mor than 1,000 mg/L dissolved solids; these supplies are used with no obvious ill effects.		
Alkalinity (as CaCO ₃)			Alkalinity is a measure of the concentration of bicarbonate and carbonate ions i water and refers to the water's ability to neutralize acid. In moderate amount alkalinity does not affect most uses of water. Its significance is usually dependent on the nature of the cations (Ca, Mg, Na, K) associated with it.		
Hardness (as CaCO ₃) NA		IA	Hardness is caused almost entirely by compounds of calcium and magnesium and is commonly recognized by the increased quantity of soap required to produce lather. Hard water is also objectionable because it causes the formation of scale in boilers, water heaters, radiators, and pipes. Total hardness can be expressed as carbonate and noncarbonate hardness. Carbonate hardness equivalent to the concentration of bicarbonate and carbonate in the water (the is, the alkalinity); any hardness in excess of this amount is called the nor carbonate hardness.		
Sodium (Na) and potassium (K)	NA		Sodium and potassium in moderate quantities have little affect on the usefulnes of the water. Water that contains more than 100 mg/L of sodium and potassium combined may cause foaming in boilers. High concentrations of sodium may be objectionable for people on a low sodium diet.		
Calcium (Ca) and magnesium (Mg)	NA		Calcium and magnesium in addition to being the main causes of hardness water are responsible for the formation of boiler scale and deposits in hot water heaters and pipes. They reduce the lathering or sudsing ability of soaps. The combine with soap to form scum deposits.		
Manganese (Mn)	-	0.05 mg/L	Manganese in concentrations as low as 0.01-0.02 mg/L causes dark brown of black stains on fabrics and porcelains and impairs the taste of coffee, tea, an other beverages.		
Nitrate (as NO ₃) (as N)	45 mg/L 10 mg/L ³		Serious and occasionally fatal poisonings have occurred when water high initrate has been used by infants. It is the cause of methemoglobinemia, ocyanosis, a sickness that gives a baby's skin a bluish tinge. Nitrate concentrations greater than 5 mg/L as NO ₃ may indicate pollution.		
Fluoride (F)	2.0 mg/L ⁴		Fluoride is desirable in water supplies in concentrations that range from 0.8 1.2 mg/L, because it reduces tooth decay. However, mottling of tooth enamel growing children in lowa may occur when fluoride concentrations exceed mg/L.		
Chloride (CI)		250 mg/L	Chloride concentrations less than 150 mg/L generally are satisfactory for mo purposes. More than 250 mg/L generally is objectionable for public supplie because chloride may combine with other ions to give a noticeable taste. Largamounts of chloride in water high in calcium and magnesium increases water corrosiveness.		
Sulfate (SO ₄)		250 mg/L	Sulfate may cause detectable taste at concentrations of 300-400 mg/L. It commonly has a laxative effect when the concentration is above 600 mg/particularly when combined with magnesium or sodium. This laxative effect commonly noted by new and casual users of the water but some become a climated to it in a short time. The effect is noticeable in almost all persons whe concentrations exceed 750 mg/L. Sulfate combined with calcium forms a har scale in boilers and water heaters.		
Bicarbonate (HCO ₃) and Carbonate (CO ₃)	NA		Bicarbonate and carbonate contribute to alkalinity of water. Bicarbonate formed when carbon dioxide in water reacts with carbonate rocks such limestone and dolomite. Small amounts of carbon dioxide are present rainwater; more is picked up by water percolating through decaying organ matter. Bicarbonate is the principal anion in much of the earth's natural frewater and ranges from about 150 to 500 mg/L in supplies from the Jord aquifer. Carbonate is seldom present in natural fresh water.		
Trace Metals Arsenic (As) Barium (Ba) Cadmium (Cd) Chromium (Cr) Lead (Pb) Mercury (Hg) Selenium (Se) Silver (Ag) Copper (Cu) Zinc (Zn)	0.05 mg/L 1.0 mg/L .010 mg/L .05 mg/L .05 mg/L .002 mg/L .01 mg/L .05 mg/L	1.0 mg/L 5.0 mg/L	Extremely small amounts of trace elements occur in water. Some of these trace metals are toxic to man. Others are essential to life, but in high concentration can be toxic or esthetically unpleasing. The full significance of these minor elements in human metabolism isn't understood in spite of much research. Regulations of the amounts that can be present in drinking water supplies have bees set by the EPA. As of this writing the water from the Jordan aquifer does not seem to have objectionable concentrations of any trace metals.		
Radionuclides Radium	5 pCi/L ⁵		The effect and significance of low concentrations of radium in public water supplies isn't fully known. However, the EPA takes the position that any dose consizing radiation has a potential to produce deleterious effects on human healt and that the effect will be proportional to the dose received. Therefore, the maximum contaminant levels established for radium concentration and gross alphactivity in community water supplies is based on the effects on human healt		

air conditioning purposes. ¹ The usage of contaminant level in this atlas follows the U.S. Environmental Protection Agency's (EPA) usage in the National Interim Primary Drinking Water Regulations (Federal Register, v. 40, no. 248 and v. 41, no. 133) and in the Proposed Secondary Drinking Water Regulations (Federal

5 pCi/L⁵

15 pCi/L

Radium

(radium-226 and

radium-228

combined)

Gross alpha

Temperature

activity

Register, v. 42, no. 62). ² NA — Not applicable.

³The maximum contaminant level for nitrate applies also to

noncommunity public water supplies.

DATA NOT SHOWN ON MAPS

aquifer vary considerably, both areally and temporally. It should be noted that iron concentration in a water supply can be affected by a number of factors, some of which tend to raise the concentration and some of which tend to lower the concentration. (A discussion of these is not within the scope of this report.) The analyses on hand indicate that the concentration of iron ranges from 0.0 to 5.6 mg/L with a mean value of 1.08 mg/L and a median value of 0.65 mg/L. Of these analyses, 70 percent exceeded the proposed secondary drinking water standard of 0.3 mg/L. The iron, however, can be easily stabilized or removed by aeration and filtration.

IRON concentrations in water samples from the Jordan

MANGANESE concentrations range from 0.0 to 0.36 mg/L with a mean value of 0.02 mg/L. The proposed maximum level of 0.05 mg/L was exceeded in only about 15 percent of the water supplies. Objectionable manganese concentrations can be lowered to about 0.05 mg/L by aeration and filtration. NITRATE concentration in water from the Jordan aquifer

⁵ The State may require annual monitoring of supplies that exceed 3 pCi/L radium-226.

throughout the State is very low. The concentration ranges

from 0.0 to 5.5 mg/L with a mean value of 1.4 mg/L. There may

be some exceptions where the Jordan aquifer is close to the

surface in northeast lowa and, thus, subject to contamination

⁴The maximum contaminant level of fluoride, which is based

on average maximum daily air temperatures, is 2 mg/L for

about 95 percent of the State it is 1.8 mg/L for a few communities in extreme southeastern and southwestern lowa,

and is 2.2 mg/L for a few communities in northeastern lowa.

that have been observed at higher doses and dose rates. The monitoring of

radioactivity in community water supplies is a screening process: when gross

alpha activity exceeds 5 pCi/L, the same or equivalent sample shall be analyzed

for radium-226; if the concentration of radium-226 exceeds 3 pCi/L, the same or

equivalent sample shall be analyzed for radium-228. The combined radium-226

and radium-228 should not exceed 5 pCi/L. Also, the gross alpha activity (includ-

ing radium-226, but excluding radon and uranium) should not exceed 15 pCi/L.

Many communities with water supplies obtained from the Jordan aquifer have

concentrations above these limits, but by zeolite, reverse osmosis, or lime-soda

ash softening treatment the radioactivity is readily removed — as much as 90

Temperature is a major consideration in use of water for industrial cooling and

percent removal is possible (Schliekelman, 1976).

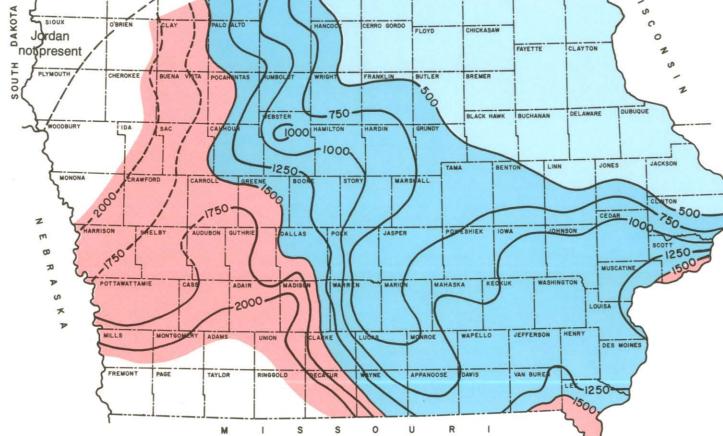
from surface sources. CHLORIDE concentration in water from the Jordan aquifer ranges from about 50 to 100 mg/L in most of the State except in the southwest quarter, the southern two tiers of counties in the southeast quarter, and in the Scott County region where concentrations range from 300 to 680 mg/L. The area where the chloride concentration exceeds 250 mg/L is shown on the sodium distribution map. HYDROGEN SULFIDE is an objectionable constituent of Jordan aquifer water supplies in some places, as at Marion,

Toledo, and Fort Dodge. Aeration treatment is used to remove

the odor before the water goes to the consumer. Quantitative

data on concentrations of hydrogen sulfide are not available at

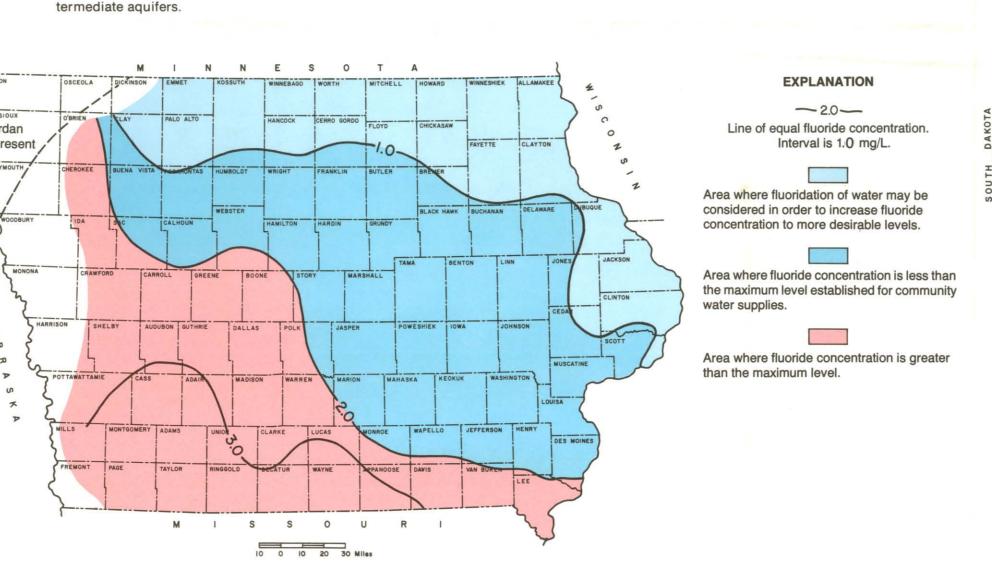
fects. Some of these supplies are the best that are locally



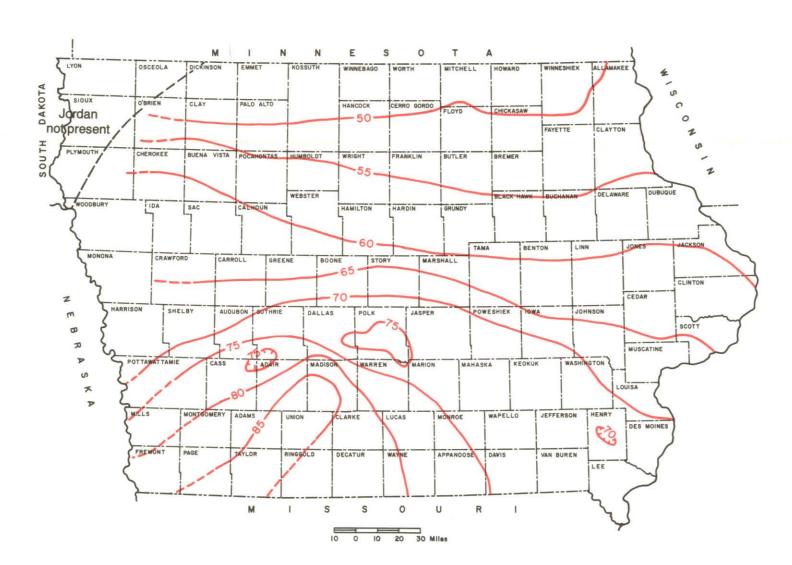
M I N N E S O T A

10 0 10 20 30 Miles ALTHOUGH THE CONCENTRATION OF DISSOLVED SOLIDS MEETS THE SECONDARY DRINKING WATER REGULATIONS IN ONLY THE NORTHEASTERN PART OF THE STATE. WATER FROM THE AQUIFER IS CONSIDERED ACCEPTABLE FOR MOST USES OVER A MUCH BROADER AREA IN SOUTHEASTERN, CENTRAL, AND NORTH-CENTRAL IOWA. If large quantities of water are needed it generally is necessary to drill to the Jordan aquifer, especially in central and southeastern lowa because only low yields and (or) highly mineralized waters are available from the upper or in-

termediate aquifers. Only locally in these areas is better quality water available from the upper or in-

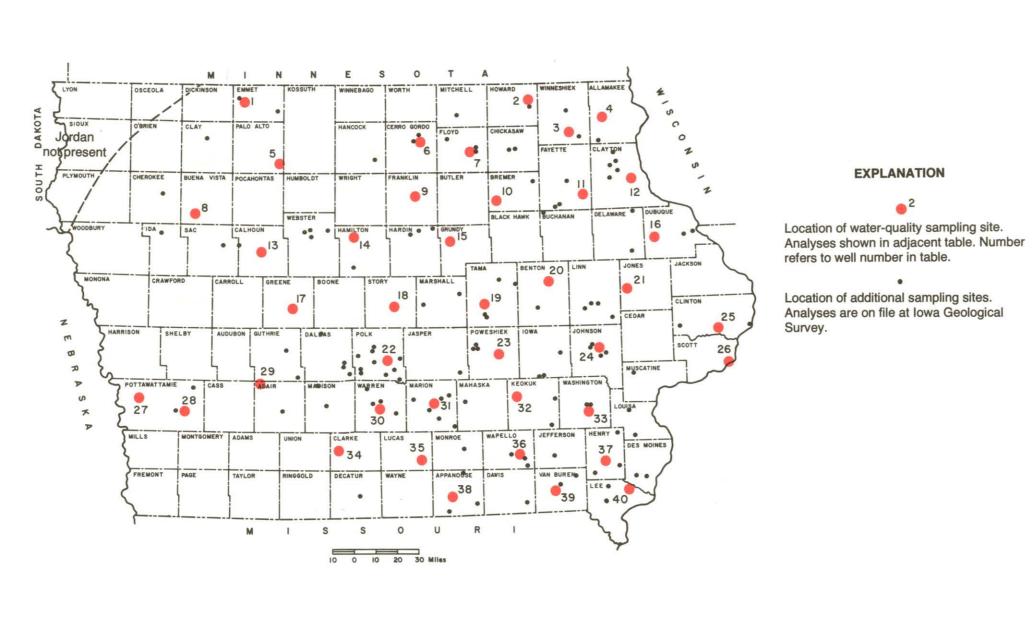


THE FLUORIDE CONCENTRATION IN WATER FROM THE JORDAN AQUIFER IS ACCEPTABLE FOR COMMUNITY WATER SUPPLIES IN A LARGE AREA OF THE STATE. Excess concentrations occur in



much of the southern and western parts of the State.

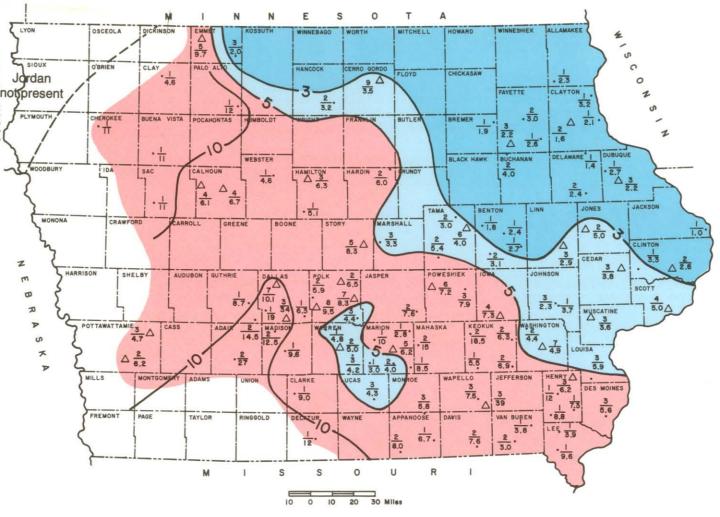
The temperature of the water obtained from the Jordan aquifer in lowa depends on the location of the wells and the depth of the aquifer, and ranges from about 50°F across the northern part of the State to about 85°F in the southwestern part.



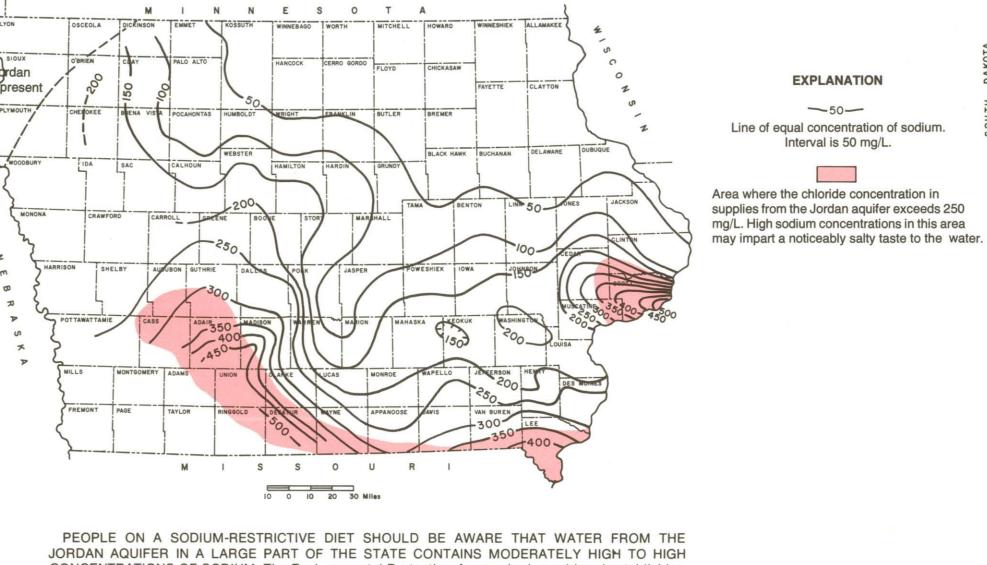
The chemical-quality data, other than radium, presented on this sheet are based on about 250 analyses of water from 185 wells that tap the Jordan aquifer. The radium-226 data are based on about 240 analyses of water from 72 municipal wells that tap the aquifer. Some of the wells are open to other water-bearing units, but the Jordan Sandstone is considered the principal water-yielding unit. Hence, if these wells are pumped several hours or longer, the water sample is considered to be representative of the Jordan aquifer. Nevertheless, the emphasis in drawing the chemical characteristic maps was placed on those analyses of water from wells that are cased from the surface into the upper part of the Prairie du Chien Formation.

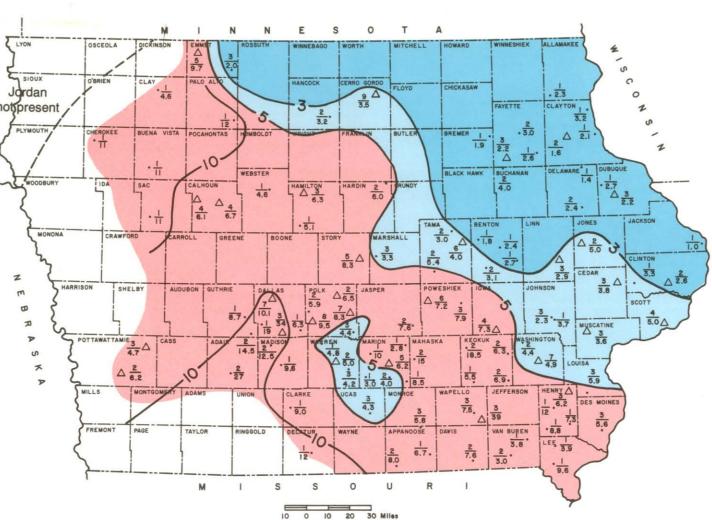
M I N N E S O T A Line of equal concentration of dissolved Area where dissolved solids are less than the Area where water from the aquifer may be considered acceptable for most uses, except concentration exceeds 2 mg/L and (or) the Area where water is of objectionable quality M I S S O U R I 10 0 10 20 30 MHes

CONCENTRATIONS OF SODIUM. The Environmental Protection Agency had considered establishing a maximum contaminant level for sodium, but concluded that "existing data did not support the establishment of a Maximum Contaminant Level for sodium in Interim Primary Drinking-Water Regulations. It is recommended that the States institute programs for regular monitoring of the sodium content of drinking water served to the public, and for informing physicians and consumers of the sodium concentrations in drinking water. By this means, those affected by high sodium concentrations can make adjustments to their diets, or seek alternative sources of water to be used for drinking and food preparation" (U.S. Environmental Protection Agency, March 31, 1977).



OBJECTIONABLE CONCENTRATIONS OF RADIUM OCCUR IN WATER FROM THE JORDAN AQUIFER IN A WIDE AREA IN SOUTHEASTERN, CENTRAL, AND WESTERN IOWA AND POSE A POTENTIAL HEALTH PROBLEM. The map above reflects the minimum concentration of radium, because only radium-226 concentrations are shown. Although the State Hygienic Laboratory has determined the radium-228 concentrations in many "finished" public-water supplies, not enough analyses have been made on "raw" water samples to use for mapping purposes. The limited data on hand indicate that radium-228 concentrations range from less than 0.68 to about 5 pCi/L or more. Consequently, the total radium concentration in water from the Jordan will likely be somewhat higher than this map indicates. Fortunately, however, high radium concentrations can be brought to





acceptable levels by softening treatment methods.

CHEMICAL ANALYSES OF WATER FROM SELECTED JORDAN WELLS

Dissolved constituents and hardness in milligrams per liter. Analyses made by the Iowa State Hygienic Laboratory, The University of Iowa

1965 St. Peter through 12-16-69 750 48 1270 1.7 .23 236 71 6.7 56 0 499 560 1.5 0.5 0.9 880 409 471 6.8 1600 6.9

Prairie du Chien- 9-26-74 | 1017 | 50 | 266 | .36 | <.01 | 58 | 26 | 2.5 | 3.0 | 0 | 281 | 41 | 1.0 | .25 | <0.1 | 240 | 230 | 10 | 7.6 | 460 | -

1965 Prairie du Chien- 10-21-69 1145 48 271 .24 < .05 72 20 1.3 2.0 0 288 38 3.0 0.9 0.4 264 236 28 7.0 470 -

1957 Prairie du Chien- 3-10-60 662 52 304 .04 <-.05 87 18 1.1 1.8 0 334 19 2.0 .05 0.8 292 274 18 7.4 530 -

1959 Prairie du Chien- 6-4-62 | 1360 | 54 | 1306 | 2.3 | <.05 | 206 | 65 | 22 | 94 | 0 | 434 | 593 | 18 | 1.0 | <0.1 | 784 | 356 | 428 | 7.2 | 1670 | -

1957 Prairie du Chien- 5-11-70 1297 50 473 .24 < .05 104 34 5.8 16 0 420 83 17 0.8 0.2 400 344 56 7.1 780 2.2

1952 Prairie du Chien-1952 Region Re

1930 Prairie du Chien- 10-1-73 | 1263 — | 407 | 2.4 | <.01 | 65 | 33 | 10 | 30 | 0 | 356 | 68 | 2 | 1.4 | <0.1 | 282 | 282 | 0 | 7.1 | 650 | —

1955 Prairie du Chien- 9-19-66 | 1312 | — | 285 | .28 | < .05 | 54 | 25 | 6.8 | 11 | 0 | 275 | 55 | 1.5 | .55 | 0.1 | 240 | 225 | 15 | 7.6 | 510 | —

1970 Prairie du Chien- 2-7-72 1970 63 1190 1.3 .03 170 60 30 100 0 375 570 10 1.3 0.4 671 307 364 6.8 1600 11.0

1972 Prairie du Chien- 1-10-77 1330 57 418 58 .01 50 26 12 69 0 310 85 23 1.0 <0.1 233 233 0 7.6 690 2.3

1928 Prairie du Chien- 12-14-60 2791 68 954 3.0 < .05 102 58 19 144 0 364 475 39 1.8 < 0.1 496 298 198 7.5 1470 6.8

1948 Galena through St. Lawrence 10-12-70 1421 — 529 .10 < .05 54 20 13 100 0 320 92 85 1.0 1.4 220 220 0 7.1 910 1.9

1966 Prairie du Chien- 12-3-68 1607 63 1790 .40 < .05 96 39 28 500 0 329 340 650 1.0 1.8 400 270 130 7.3 3100 —

1956 Prairie du Chien- 7-14-75 | 2519 | 77 | 863 | 1.4 | <.01 | 61 | 26 | 19 | 200 | 0 | 267 | 300 | 98 | 3.4 | 5.5 | 256 | 219 | 37 | 8.0 | 1400 | 5.7

Prairie du Chien- 4-30-73 1765 — 1330 43 <.01 74 33 18 320 0 298 500 220 2.8 0.8 308 244 64 7.6 2100 3.2

... 1969 Prairie du Chien- 11-19-69 2520 70 1610 2.0 .10 216 54 40 220 0 200 740 210 2.4 3.2 760 164 596 7.2 2200 —

Grundy County SW NW sec. 14, T. 88N., R. 18W. 1959 Prairie du Chien- Jordan 8-6-75 2050 — 601 1.0 .01 81 38 22 71 0 382 200 13 1.3 0.4 344 313 31 7.4 950 6.2

Greene County
NW SW sec. 8, T. 83N., R. 30W.

1951

Prairie du ChienJordan

8-3-51

2307

— 1650

0.3

0 157

63

269

0 259

838

107

2.8

0 651

212

439

7.5

2020

—

Tama County SE SW sec. 15, T. 83N., R. 15W. 1961 Prairie du Chien-Jordan 2-3-72 2016 — 704 2.2 .02 96 45 19 79 0 378 260 8 1.1 <0.1 420 310 110 6.9 1000 5.2

Benton County
SE SW sec. 16, T. 85N., R. 10W.

1932

Prairie du ChienJordan

5-6-68

1505

- 568

.80

<.05

74

38

15

70

0

364

180

7

1.1

0.2

340

298

42

7.2

880

2.4

Jones County
NE NW sec. 2, T. 84N., R. 4W.

1970

Prairie du ChienJordan

4-21-71

1640

63

575

.32

.05

70

28

15

80

0

339

160

11

1.2

0.5

292

278

14

7.0

830

4.6

 Polk County NE SE sec. 13, T. 79N., R. 23W.
 1968
 Prairie du Chien-Jordan
 10-25-71
 2505
 —
 594
 5.6
 .07
 70
 29
 15
 82
 0
 305
 220
 20
 1.7
 <0.1</th>
 296
 250
 46
 7.5
 910
 6.4

 Poweshiek County NW SW sec. 23, T. 80N., R. 14W.
 1961
 Prairie du Chien-Jordan
 8-14-72
 2040
 70
 899
 3.2
 .03
 97
 48
 19
 110
 0
 364
 380
 20
 1.2
 0.2
 450
 298
 152
 7.1
 1300
 4.8

Johnson County
SW SW sec. 32, T. 80N., R. 6W.

1965
Prairie du ChienJordan
7-21-72
1677

- 1170
96
02
100
50
19
170
0
293
570
42
1.4
<-0.1
460
240
220
7.2
1600
2.4

Pottawattamie County NW NE sec. 1, T. 75N., R. 40W. 1971 Prairie du Chien-Jordan 10-7-71 2925 76 1680 .84 .02 196 61 38 240 0 203 740 220 3.0 3.9 742 166 576 7.3 2300 —

Adair County NW NE sec. 4, T. 77N., R. 33W. 1968 Prairie du Chien-Jordan 5-26-69 2700 72 1870 4.4 .06 184 58 51 330 0 207 770 320 3.1 0.5 700 170 530 7.3 2700 —

Warren County
NW NE sec. 30, T. 76N., R. 23W.

Marion County
SE NW sec. 31, T. 76N., R. 19W.

Prairie du ChienJordan

9-16-74

2590

79

624

39

<-0.01

56

25

16

130

0

317

230

32

1.9

<-0.1

236

236

0

7.4

1000

4.7

1000

4.7

1000

5.9

 Keokuk County NW SW sec. 10, T. 76N., R. 13W.
 1964
 Prairie du Chien-Jordan
 12-19-74
 1925
 —
 974
 2.5
 <.01</th>
 100
 53
 21
 140
 0
 339
 450
 33
 1.5
 1.0
 440
 278
 162
 7.4
 1400
 20

 Washington County SE NW sec. 17, T. 75N., R. 7W.
 1969
 Prairie du Chien-Jordan
 6-19-72
 1825
 —
 1200
 .27
 .01
 110
 47
 22
 180
 0
 290
 580
 60
 1.3
 <0.1</td>
 480
 238
 242
 7.2
 1700
 3.5

Clarke County
SE SW sec. 10, T. 72N., R. 27W.

1968 Prairie du ChienJordan

10-1-69 3060 82 1680 4.5 < .05 140 41 27 340 0 257 760 220 2.4 0.5 520 211 309 7.3 2500 9.0

Wapello County
NW SE sec. 36, T. 72N., R. 13W.

1968 Prairie du ChienJordan

10-21-74 2108 — 1160 2.5 < .01 86 39 20 240 0 310 500 120 1.7 1.2 382 254 128 7.8 1700 9.0

Henry County
NW NE sec. 9, T. 71N., R. 6W.

1946 Prairie du ChienJordan

5-4-67 1860 70 1120 .15 < .05 108 39 17 190 0 327 470 90 1.6 5.3 431 268 163 7.6 1700 —

Van Buren County
SE NW sec. 36, T. 69N., R. 10W.

1954

St. Peter through
Jordan

12-17-73

1831

— 1240

1.8 < .01

69

32

16

320

0

320

430

240

2.4

3.0

296

262

34

7.6

2000

2.3

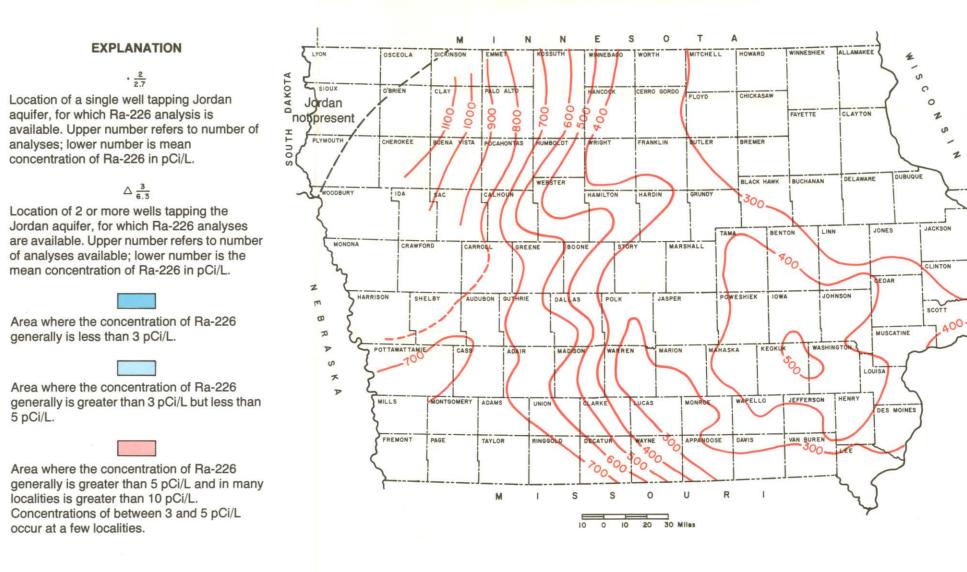
SE NW sec. 36, T. 69N., R. 10W.

Prairie du Chien- 6-10-55 | 1355 | 52 | 244 | .9 | 0 | 53 | 23 | 4.0 | 5.5 | — | 243 | 17 | 2.0 | 1.0 | 0 | 227 | 227 | 0 | 8.3 | 429 | —

Prairie du Chien- 7-17-75 | 1690 | — | 1970 | 1.7 | .03 | 290 | 74 | 45 | 160 | 0 | 323 | 1100 | 30 | 1.6 | 1.3 | 1010 | 265 | 745 | 7.2 | 2300 | 11.0

Prairie du Chien- 12-1-70 815 52 290 48 < .05 70 25 1.0 3.4 0 332 27 1.0 0.2 0.4 280 272 8 7.2 520 2.1

St. Peter through 3-31-71 2005 59 960 .68 < .05 116 41 20 130 0 356 370 65 1.2 1.1 460 292 168 6.9 1400 —



MINNESOTA

M I S S O U

cipal anion.

10 0 10 20 30 Miles

THE SULFATE CONCENTRATION IS LESS THAN THE PROPOSED MAXIMUM LEVEL FOR COM-

MUNITY WATER SUPPLIES IN ONLY ABOUT 30 PERCENT OF THE STATE. However, because most

users become acclimated to higher sulfate concentrations, the water is considered acceptable in an

additional 30 percent of the State. Note that the distribution of sulfate in the aquifer closely parallels

the dissolved-solids distribution; the major exception is in Scott County where chloride is the prin-

EXPLANATION

-50-

Interval is 50 mg/L.

generally is less than 3 pCi/L.

occur at a few localities.

WATER FROM THE JORDAN AQUIFER RANGES FROM LESS THAN 300 TO MORE THAN 1000 MG/L HARDNESS. THE LOWEST CONCENTRATIONS ARE FOUND IN THE EASTERN HALF OF THE STATE AND THE HIGHEST IN THE WESTERN HALF. The hardness of the water can be reduced by softening methods. However, it should be noted that the sodium content will increase perceptibly when an ion exchange method of softening is used.

WATER TREATMENT

ONLY A FEW OF THE JORDAN SUPPLIES ARE NOT TREATED IN SOME WAY. The majority of the Jordan aquifer well water supplies used for municipal purposes contain excess iron and require treatment for iron removal. A typical treatment system consists of aeration, filtration, and disinfection, in that order. The aerators generally are forced draft, pressure, overflow trays or other splash units. Filtration is through sand by pressure or rapid gravity flow. Chlorine gas or hypochlorine solutions are the usual disinfecting agents. About 75 to 80 percent of all municipal wells that tap the Jordan aquifer are treated in this fashion.

A relatively small number of Jordan aquifer supplies are softened as a part of their treatment process. Most of the softened supplies are located

in the southeast quarter of the State. The most common softening

method is the zeolite or base exchange treatment. This method is preferred for community supplies where radioactivity concentrations exceed 5.0 pCi/L radium-226 and radium-228 combined or the gross alpha concentration is more than 15 pCi/L. A few large plants use the lime-soda ash softening process and one or two the reverse osmosis method. Fluoridation is included in several treatment systems in the northern

EXPLANATION

-200-

Line of equal sulfate concentration.

Area where the sulfate concentration is less

Area where sulfate concentration is more

than the maximum recommended level but

Area where sulfate concentration is high

enough to affect the taste of water and also to

EXPLANATION

-300-

Line of equal hardness.

Interval is 100 mg/L.

may be considered acceptable because most

than the maximum level proposed for

community water supplies.

users can adjust to it.

have a laxative effect.

Interval is 100 mg/L.

aquifer water is less than 1.0 mg/L. In about 25 percent of the Jordan aquifer municipal supplies polyphosphate or alkali chemicals are added to the water for stabilization, to adjust the pH, for corrosion control, and to hold iron in solution. A few communities aerate the water to remove hydrogen sulfide odor.

part of the State where the natural fluoride concentration of the Jordan

THE CHEMICAL-QUALITY CHARACTERISTICS OF WATER IN THE JORDAN AQUIFER ARE A FUNCTION OF THE MINERAL COMPOSITION OF THE ROCKS THROUGH WHICH THE WATER PERCOLATES IN THE INTAKE AREA AND WHERE IN THE FLOW SYSTEM THE WATER OC-CURS. The best quality water occurs in northeast lowa, where the concentration of dissolved solids is less than 500 mg/L. The water is a calcium-magnesium-bicarbonate type that is typical of flush-through systems in predominantly carbonate terrains. The aquifer is recharged in northeastern lowa and southeastern Minnesota by vertical leakage through drift and carbonate rocks. From there, the water moves toward and discharges into the Mississippi River north of the Jackson-Clinton County line (approximately where the Ordovician Maquoketa Shale crops out in the river channel). The chemical composition of the water remains essentially the same from recharge to discharge areas, indicating the system has been thoroughly flushed.

The mineralization of the water increases markedly southwest of a line running from Hancock to Clinton Counties. Water that is classed as good-to-fair quality and containing dissolved-solids concentrations of 500 to 1500 mg/L is found in a broad belt through the central and southeastern parts of the State. From east to west in the northern half of this area, the water changes from a calcium-magnesium bicarbonate type to a calcium-magnesium sulfate-bicarbonate type. In the southern half of the area — the down-gradient direction — the sodium and chloride concentrations increase perceptibly until near the southern and southeastern borders of the State the water is a sodium-calciummagnesium sulfate-chloride-bicarbonate type. This central part of the flow system is being flushed more slowly than in northeastern lowa as Missouri and western Illinois.

SUMMARY OF QUALITY West and southwest of this central belt, the water generally is so highly mineralized as to be a poor to objectionable source for most uses. The water is a calcium-magnesium sulfate-bicarbonate type in the northern part of this area and gradually changes to a calcium-magnesium-sodium

sulfate-chloride-bicarbonate type in the southern part. In this western

part of the flow system where the transmissivity of the aquifer is lowest,

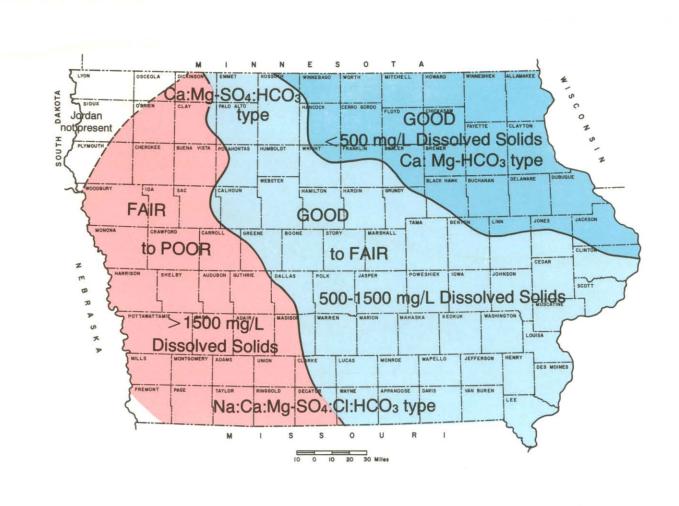
the aquifer is flushed very slowly as the water discharges by underflow into northern Missouri. The increasing sulfate concentrations from east to west in the western and central belt are attributed to recharge to the aquifer by vertical leakage through the Cretaceous shales and sandstones that locally contain water with high sulfate concentrations in northwestern lowa and south-central Minnesota.

The radium concentration also has an important bearing on the general acceptance of Jordan aquifer water supplies. The data seem to indicate a correlation between the concentration of radioactive isotopes in the Jordan aquifer and the extent to which the aquifer has been flushed. Low radium values occur in northeast lowa and higher values in

central and western lowa.

The map below illustrates the general quality of the water in the Jordan FUTURE CHEMICAL-QUALITY CHANGES CAN BE EXPECTED IN THE CENTRAL PART OF THE STATE. Poor-quality water from the western part of the flow system is being diverted toward the major pumping centers as evidenced by the potentiometric map. Because the flow velocities are extremely slow, the change will be gradual. Significant chemical quality changes are not expected in the northeastern part of the State. although the Allamakee-Clayton-Winneshiek County area is susceptible

to nitrate and bacterial pollution from surface-water infiltration.



PAUL J. HORICK **Iowa Geological Survey** WALTER L. STEINHILBER Published by the STATE OF IOWA 1978



United States Geological Survey